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Changes in Membrane Permeability During Rehydration of Freeze-Dried Weissella paramesenteroides LC11

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Abstract: The objective of this study was to assess changes in membrane permeability during rehydration of freeze-dried *Weissella paramesenteroides* LC11. Viability was assessed using the electrical conductivity measurement (ms/cm/g dry weight) and the plate count method (cfu g⁻¹ dry weight). The symptoms of injury included an increase in the electrolyte leakage during the first 4 h of rehydration in Milli Q water and a decrease in the survival rate, suggesting an increase in membrane permeability during dehydration. An increase in the temperature, NaCl or monosodium glutamate concentration and a decrease in H⁺ concentration resulted in an increase in the electrolyte leakage and a decrease in the survival rate. However, a decrease in the electrolyte leakage was observed with increasing glycerol, sucrose or maltodextrin and resulted in the maintenance of cell viability. Regression analysis showed that overall, there was a good correlation between the survival rate and electrolyte leakage (R²>0.90). Change in membrane permeability might lead to electrolyte leakage during rehydration and ultimately, cell death.

Key words: Viability, starter culture, membrane integrity, freeze-drying, lactic acid bacteria, fermented foods

INTRODUCTION

Freeze drying is commonly used to produce concentrated Lactic Acid Bacteria (LAB) starter cultures for the food industry (Manel *et al.*, 2009). After dehydration, one population of microorganisms might be killed, another population might survive (non-injured) and a third might be sub-lethally injured. It is now generally agreed that the cell membrane is the component most commonly affected after a sub-lethal injury. Santivarangkna *et al.* (2007) and Brennan *et al.* (1986) indicated that a substantial alteration of membrane structure occurs in dried *Lactobacillus Helveticus* and *Lb. acidophilus*, respectively.

These researchers indicated that leakage of the cell membrane was indirectly evident by an increase in cellular components such as enzymes and UV-absorbing materials in the surrounding medium after rehydration of dried cells, or by an increase in the cells' sensitivity to chemicals such as NaCl, oxgall and lysozyme. To evaluate dried bacteria resistance to dehydration and control changes in membrane permeability through the use of additives, it

would be necessary to conduct a quick and inexpensive method immediately after the drying process. Most studies performed to determine the efficiency of conservation methods are based on the survival rate of the dried bacteria. However, cell viability can also be determined bv measuring the electrochemical characteristics of the culture medium, which are related to both the number and the metabolic activity of the microorganisms (Lanzanova et al., 1993). The electrolyte leakage assay was used to evaluate the performance of freeze-dried Lb. delbrueckii sp. bulgaricus and LAB in milk (Martos et al., 1998; Lanzanova et al., 1993). Such measurements, however have never been performed in combination with the survival rate to assess the change in membrane permeability during rehydration of freeze dried LAB.

In this research, the viability of the freeze-dried *W. paramesenteroides* was evaluated by measuring electrolyte leakage and viable cell after the drying process. The effects of the pH, temperature, NaCl or glycerol concentration and sucrose, maltodextrin or monosodium glutamate concentration during rehydration

of the freeze-dried strain were investigated. Previously, *W. paramesenteroides* LC11 had been isolated during cassava fermentation and was selected as a suitable starter culture for gari production (Yao *et al.*, 2008, 2009; Kostinek *et al.*, 2007).

MATERIALS AND METHODS

Microorganism, production and freeze-dying: The LAB W. paramesenteroides LC11 was provided by the Federal Research Centre for Nutrition, Institute of Hygiene and Toxicology (Karlsruhe, Germany), produced in a 20 L bioreactor containing MRS medium for 18 h, concentrated and then freeze-dried as described by Yao et al. (2008).

Experimental procedure: In order to determine the time course of electrolyte leakage during rehydration, 0.5 g of the dried sample (original concentration about 4.8×10¹² cfu g⁻¹ dry weight) was soaked in 10 mL Milli Q water for 8 h at room temperature. Samples were sampled every 2 h and analyzed for conductivity and viability. For the next steps of the experimental procedure 0.5 g was soaked in 10 mL Milli Q water for 2 h and then transferred into 10 mL Milli Q water for 4 h that had been obtained according to the designed parameter. The effect of each parameter was evaluated through conductivity and viable counts determinations in 20 mL total Milli Q water after 6 h (total experimental time) at room temperature. The effect of temperature was studied from 20-40°C at 5°C intervals. The effect of pH was studied in a 10 mL buffer solution (0.1 M KH₂PO₄ solution adjusted with 0.1 N NaOH) at pH 5.8, 6.2, 6.6, 7.0 and 7.4. The effect of glycerol, NaCl, sucrose, maltodextrin or monosodium glutamate concentration was studied at 5, 15, 20, 30 and 40 g L⁻¹. All the experiments were carried out aseptically 20 h after the freeze-drying process. Maltodextrin IT 12 or glycerol and monosodium glutamate or all other chemicals were purchased from Roquette (Lestren, France) and Sigma-Aldrich (Schnelldorf, Germany), respectively.

Analyses: The electrical conductivity (ms/cm/g DW) of the solution was measured using a ProfiLine Multi 197i (WTW GmbH, Weilheim, Germany) conductivity meter. Total conductivity was obtained after autoclaving the samples at 112°C for 20 min. The viable counts were obtained using the plate count method after 48 h at 30°C with MRS-agar medium (pH 6.0). The dry cell weight was determined after drying at 105°C until constant weight.

Data analysis: The percentage leakage was calculated using the ratio (EC_f:EC_i)/(EC_{total}-EC_i)×100. The percentage

leakage was expressed as percentage of total electrolytes after autoclaving the samples for this reason; after autoclaving, we expect that cells die that means at first stage a decrease of the cells' membrane electrochemical potential to zero, which will result in a transient decrease of the total conductivity (dead cells) and ultimately to a full leakage of intracellular metabolites to the medium compared to the initial conductivity (i.e., experimental conductivity) (cells alive). The percentage survival was calculated as N/N₀×100. A regression analysis was made between the percentage leakage and the percentage survival and the goodness-of-fit was confirmed by the R². Productions in the bioreactor were done in duplicate. The data were compared using Tukey's honestly significant difference (Statistica 7.1, StatSoft Inc., 2005). The level of significance was set at p<0.05.

RESULTS AND DISCUSSION

The leakage of electrolytes from the freeze-dried W. paramesenteroides bacteria during rehydration in Milli Q water was a function of time (Fig. 1a). During the rehydration period, the percentage leakage increased significantly between the 2nd and 4th h (from about 8-14% of total electrolytes) and then remained fairly constant up to 8 h (p<0.05). The viable population of the freeze-dried strain was reduced from an initial population of 4.8×10^{12} cfu g⁻¹ DW (viable count at the end of freeze drying) to 1.8×10¹² cfu g⁻¹ DW with a percentage survival of 36±5.4% after 4 h of rehydration. More than 90% of the variability in the percentage survival could be explained by the percentage leakage, as shown by the coefficient of determination of the regression lines ($R^2 = 0.9$). However, the change in the percentage leakage for each unit change in the percentage survival is not significant as shown by the slopes of the lines which vary from 0.2-2.3. The electrolyte leakage assay can not be use alone as a predictor of cell viability. The research suggests that 4 h was an appropriate time for an accurate measurement of electrolyte leakage from dried cells, since beyond this time there was no further increase in electrolytes efflux. In this way, the method became less time-consuming for routine work. The increase in the percentage leakage indicated that changes in membrane permeability of the dried strain might be induced by the rehydration treatment. It is possible that loss of membrane structure could lead to electrolyte leakage during rehydration and ultimately, cell death. The findings accord with the conclusion reached by others researchers who showed that membrane damage was a catalyst of Lb. acidophilus death during drying (Brennan et al., 1986). Because of an increase in cell rupture as a result of dehydration treatment the

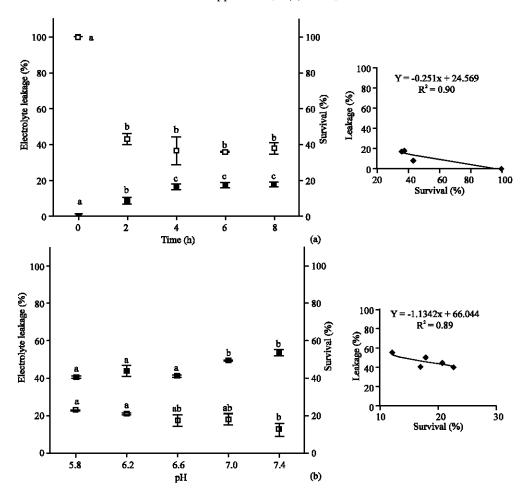


Fig. 1: Time course of (a) or Effect of external pH on (b) electrolyte leakage (■) and survival (□) during rehydration of freeze-dried *W. paramesenteroides* in Milli Q water. Values are presented as means±SD (n = 4). Mean values with a different letter for each parameter are significantly different (Tukey HSD test, p<0.05)

freeze-dried cells were incubated for 2 h before experiments. The percentage leakage or survival increased or decreased with decreasing H+ concentration in the solution, respectively (Fig. 1b). For example, the percentage leakage increased significantly from pH 6.6-7.4 by 13% of total electrolytes with a decrease in the viable population of about 5% (p<0.05). The dehydration stress could increase the rate of passive diffusion across the phospholipid bilayer. In cell membranes, K⁺ and Na⁺ are believed to be transported across the plasma membrane by an ionic pump which transports K⁺ into the cell and Na⁺ out of it, in both cases, against its electrochemical gradient (Yeagle, 1993). This results in a difference in chemical potential across the membrane, which represents a delicate balance in selective permeability and concentration gradients. An increase in membrane permeability might lead to a decrease in electrical charge across the membrane and ultimately, cell death.

Alternatively, an increase in electrolyte leakage would occur if the activity or the efficiency of the transport system had been altered. Whether or not these changes are related to the freeze-drying process remains to be established. The percentage leakage significantly from 20-40°C by 13% of total electrolytes, with a decrease in viable population of about 32% (Fig. 2a) (p<0.05). An increase in the percentage leakage, with an increase in the rehydration temperature, would occur if the integrity of the membrane was altered. An increase in NaCl or glycerol concentrations (5-40 g L⁻¹) was associated with a decrease in osmotic pressure from -3.6 to -4.2 Pa or from -3.2 to -5.3 Pa, respectively. The osmotic pressure data given are only approximate, since they do not take into consideration those of the dried powder. The percentage leakage increased with increasing NaCl concentration and could indicate that electrolytes efflux might be imposed by the

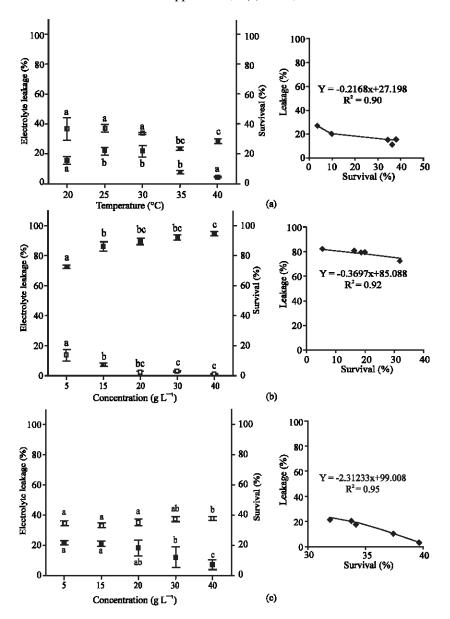


Fig. 2: Effect of temperature (a) and NaCl (b) or glycerol (c) concentration on electrolyte leakage (■) and survival (□) during rehydration of freeze-dried *W. paramesenteroides*. Values are presented as means±SD (n = 4). Mean values with a different letter for each parameter are significantly different (Tukey HSD test, p<0.05)

osmotic gradient (Fig. 2b). While an increase in NaCl concentration was associated with a decrease in viability, that of glycerol resulted in the maintenance of cell viability or a better viability (Fig. 2c). For example, 7% leakage and 40% survival rate were achieved with 40 g L⁻¹ glycerol, compared with 81% leakage and 3% survival rate with 40 g L⁻¹ NaCl, after 4 h of rehydration. The sensitivity of the freeze dried *W. paramesenteroides* to an increase in NaCl concentration could also indicate damage to cell membrane components. It is possible that the

presence of glycerol makes the bacterial membrane less permeable and therefore, it might improve the viability. Figure 3a and b shows that the percentage leakage decreased as the concentration of sucrose or maltodextrin increased (from 5-40 g L⁻¹), respectively. At 40 g L⁻¹ concentration, a higher percentage survival and a lower percentage leakage were obtained with maltodextrin (48 and 1%, respectively), followed by sucrose (45 and 4%, respectively). These results are consistent with those obtained above on the effect of an increase in glycerol

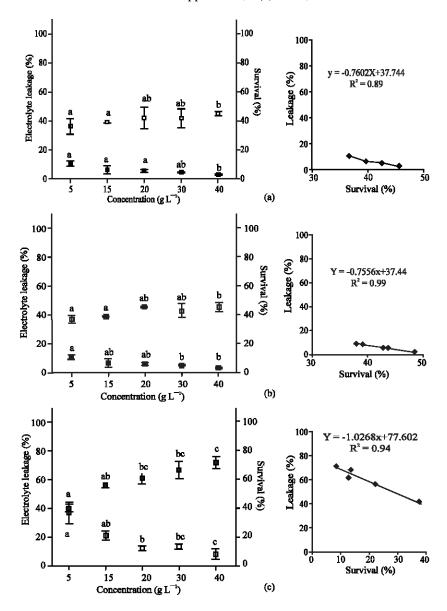


Fig. 3: Effect of sucrose (a), maltodextrin (b) and monosodium glutamate (c) concentration on electrolyte leakage (■) and survival (□) during rehydration of freeze-dried *W. paramesenteroides*. Values are presented as means±SD (n = 4). Mean values with a different letter for each parameter are significantly different (Tukey HSD test, p<0.05)

concentration. In contrast, an increase in monosodium glutamate concentration resulted in an increase in the percentage leakage and a decrease in viability in the same range of concentration (Fig. 3c). Sucrose, maltodextrin and monosodium glutamate are commonly used as protective compounds in the freeze drying process (Oldenhof *et al.*, 2005).

The results suggest that increasing the number of OH groups in the rehydration solution might make the membrane less permeable to electrolytes and could

improve or maintain the viability, possibly by protecting membrane phospholipids. These results accord with those reported by other researchers, who have shown that the direct interaction between sucrose and membrane might contribute to the protective effects of sucrose on the dried cells of *Lb. Bulgaricus* (Oldenhof *et al.*, 2005). These researchers suggested that OH groups of sugar and/or polyalcohol could replace water and could interact with phospholipids to maintain membrane integrity.

CONCLUSION

Changes in membrane permeability of the freeze-dried *W. paramesenteroides* was assessed by measuring electrolyte leakage and viable cells during rehydration. An increase in the leakage of solutes from the dried strain was correlated with a decrease in viable cell during rehydration. The dehydration process might increased the incidence of cell rupture during rehydration. The electrolyte leakage assay associated with the plate count method, could be used as a predictive test for the description of dehydration inactivation of dried bacteria.

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